

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1068). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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2-(5-Nitro-2-styryl-1-imidazolyl)ethanol

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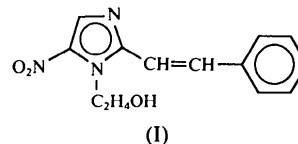
Abstract

The molecules of the title compound, $C_{13}H_{13}N_3O_3$, are linked in chains through hydrogen bonds along the *a* direction.

Comment

The perspective view of the title compound (I) showing the atom-numbering scheme and hydrogen bonding is given in Fig. 1. The crystal contains well ordered molecules of 2-(5-nitro-2-styryl-1-imidazolyl)ethanol in the form previously observed in crystals of 2-(2-methyl-5-nitro-1-imidazolyl)ethanol (metronidazole) (Blaton, Peeters & De Ranter, 1979). The bond lengths and angles are in good agreement to within 0.005 Å and 1.6°, respectively, with those observed in metronidazole. More significant deviations are observed for C2—N3 (longer by

0.012 Å) and C2—C21 (shorter by 0.027 Å). These differences are undeniably a result of the conjugation effect between the C2—C21 single bond and the neighbouring C2=N3 and C21=C22 double bonds.



The five-membered ring is planar within 0.003 (2) Å. The nitro-group plane crosses the weighted least-squares imidazole plane at an angle of 4.7 (1)°. The benzene ring is planar and its weighted least-squares plane forms a dihedral angle of 5.7 (1)° with the imidazole plane. Although the molecule was expected to be planar except for the ethanol moiety, twisting is observed around the C2—C21 bond [N3—C2—C21—C22 21.9 (3)°]. This is probably caused by the molecular crowding or packing effects.

The molecules of the title compound are linked through O13—H13...N3' hydrogen bonds [1.99 (3) Å, 166 (2)°; symmetry code: (i) $x - 1, y, z$] forming chains along the *a* direction. Although for both the title compound and metronidazole (Blaton *et al.*, 1979) the same atoms are engaged in hydrogen bonding, the hydrogen-bonding patterns are different. This results in significant differences in the orientation of the ethanol moiety. The values of the torsion angle C5—N1—C11—C12 for the title molecule and metronidazole (Blaton *et al.*, 1979) are -106.8 (2)° and 82.6 (2)°, respectively.

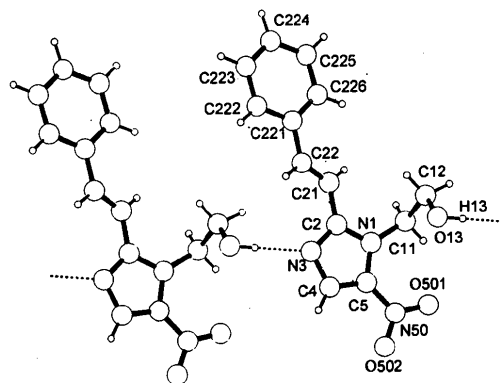


Fig. 1. The atomic numbering scheme and hydrogen bonding.

Experimental

Crystal data

$C_{13}H_{13}N_3O_3$
 $M_r = 259.26$
Monoclinic
 $P2_1/c$
 $a = 7.088$ (1) Å
 $b = 7.116$ (1) Å
 $c = 24.935$ (3) Å
 $\beta = 95.03$ (1)°

Mo $K\alpha$ radiation
 $\lambda = 0.71069$ Å
Cell parameters from 40
reflections
 $\theta = 10.38$ – 12.48°
 $\mu = 0.094$ mm⁻¹
 $T = 293$ K
Prism

$V = 1252.8 (3) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.375 \text{ Mg m}^{-3}$

$0.30 \times 0.30 \times 0.20 \text{ mm}$
 Light yellow

N1—C2—C21	122.8 (1)	N1—C5—C4	107.6 (2)
C21—C2—N3	125.6 (2)	C4—C5—N50	126.2 (2)
C2—C21—C22	123.5 (2)	N1—C5—N50	126.2 (1)
C21—C22—C221	125.4 (2)	C5—N50—O502	116.5 (2)
C22—C221—C226	122.8 (2)	C5—N50—O501	120.6 (2)
C22—C221—C222	119.5 (2)	O501—N50—O502	122.8 (2)
C222—C221—C226	117.6 (2)		

Data collection

Stoe Stadi-4 four-circle diffractometer
 $R_{\text{int}} = 0.012$
 $\theta_{\text{max}} = 27.5^\circ$
 ω scans
 $h = -9 \rightarrow 9$
 Absorption correction: none
 $k = -9 \rightarrow 9$
 5956 measured reflections
 $l = 0 \rightarrow 32$
 2879 independent reflections
 3 standard reflections
 2014 observed reflections
 frequency: 60 min
 $|I| > 2.0\sigma(I)$
 intensity variation: 3.0%

Refinement

Refinement on F
 $R = 0.0361$
 $wR = 0.0577$
 $S = 0.970$
 2014 reflections
 224 parameters
 All H-atom parameters refined
 $w = 1/[\sigma^2(F) + 0.003F^2]$

$(\Delta/\sigma)_{\text{max}} = 0.003$
 $\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$
 Extinction correction: none
 Atomic scattering factors from *CRYSRULER* (Rizzoli, Sangermano, Calestani & Andreetti, 1989)

Data collection: *DIF4* (Stoe & Cie, 1992a). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1992b). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELXL76* (Sheldrick, 1976). Molecular graphics: *PLUTON93* (Spek, 1993). Software used to prepare material for publication: *PARST* (Nardelli, 1983).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1069). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$$

	x	y	z	U_{eq}
N1	0.4152 (2)	0.1911 (2)	0.0450 (1)	0.0327 (3)
C11	0.2481 (2)	0.1100 (2)	0.0676 (1)	0.0405 (5)
C12	0.1434 (2)	0.2555 (3)	0.0977 (1)	0.0473 (5)
O13	0.0892 (2)	0.4119 (2)	0.0656 (1)	0.0563 (4)
C2	0.5749 (2)	0.2478 (2)	0.0748 (1)	0.0336 (4)
C21	0.6021 (2)	0.2243 (2)	0.1328 (1)	0.0383 (4)
C22	0.7256 (2)	0.3236 (2)	0.1644 (1)	0.0391 (4)
C221	0.7616 (2)	0.3009 (2)	0.2229 (1)	0.0398 (4)
C222	0.9276 (3)	0.3713 (2)	0.2490 (1)	0.0515 (6)
C223	0.9714 (3)	0.3451 (3)	0.3036 (1)	0.0618 (7)
C224	0.8482 (3)	0.2507 (3)	0.3334 (1)	0.0602 (6)
C225	0.6807 (3)	0.1840 (3)	0.3089 (1)	0.0631 (7)
C226	0.6366 (3)	0.2078 (3)	0.2541 (1)	0.0557 (6)
N3	0.7010 (2)	0.3226 (2)	0.0438 (1)	0.0375 (3)
C4	0.6206 (2)	0.3132 (2)	-0.0072 (1)	0.0378 (4)
C5	0.4451 (2)	0.2342 (2)	-0.0075 (1)	0.0342 (4)
N50	0.3158 (2)	0.2043 (2)	-0.0534 (1)	0.0438 (4)
O501	0.1575 (2)	0.1436 (3)	-0.0486 (1)	0.0761 (6)
O502	0.3727 (2)	0.2415 (2)	-0.0972 (1)	0.0678 (5)

Table 2. Selected geometric parameters (\AA , $^\circ$)

N1—C11	1.473 (2)	C221—C226	1.396 (3)
N1—C2	1.360 (2)	C222—C223	1.382 (3)
N1—C5	1.379 (2)	C223—C224	1.371 (3)
C11—C12	1.511 (2)	C224—C225	1.371 (3)
C12—O13	1.405 (2)	C225—C226	1.385 (3)
O13—H13	0.86 (2)	N3—C4	1.349 (2)
C2—C21	1.452 (3)	C4—C5	1.365 (2)
C2—N3	1.342 (2)	C5—N50	1.418 (2)
C21—C22	1.329 (2)	N50—O501	1.218 (2)
C22—C221	1.468 (3)	N50—O502	1.226 (2)
C221—C222	1.387 (2)		
C2—N1—C5	105.2 (1)	C221—C222—C223	121.4 (2)
C11—N1—C5	130.2 (1)	C222—C223—C224	120.1 (2)
C11—N1—C2	124.5 (1)	C223—C224—C225	119.7 (2)
N1—C11—C12	111.6 (1)	C224—C225—C226	120.6 (2)
C11—C12—O13	112.4 (2)	C221—C222—C225	120.6 (2)
C12—O13—H13	107 (2)	C2—N3—C4	106.0 (1)
N1—C2—N3	111.6 (1)	N3—C4—C5	109.6 (2)

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Isopropyl Hydrogen 9,10-Dihydro-9,10-ethenoanthracene-11,12-dicarboxylate

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Abstract

The title compound, C₂₁H₁₈O₄, exists in the crystal as hydrogen-bonded dimers, O—H...O = 2.669 (3) \AA . The molecule has normal dimensions. The C=O bond